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Waterlike hierarchy of anomalies in a continuous spherical shouldered potential

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We investigate by molecular dynamics simulations a continuous isotropic core-softened potential with attractive well in three dimensions, introduced by Franzese [J. Mol. Liq. 136, 267 (2007)], that displays liquid-liquid coexistence with a critical point and waterlike density anomaly. Besides the thermodynamic anomalies, here we find diffusion and structural anomalies. The anomalies, not observed in the discrete version of this model, occur with the same hierarchy that characterizes water. We discuss the differences in the anomalous behavior of the continuous and the discrete model in the framework of the excess entropy, calculated within the pair correlation approximation. © 2008 American Institute of Physics. [DOI: 10.1063/1.2830706]

I. INTRODUCTION

Experiments for water allow us to locate the line of temperatures of maximum density (TMD) below which the density decreases with decreasing temperature at constant pressure, instead of increasing as in the majority of fluids. Experiments for Te, 2 Ga, Bi, 3 S, 4,5 and Ge $_{15}$ Te $_{85}$, 6 and simulations for silica, $^{7-10}$ silicon, 11 and BeF $_2$, 7 show, as well, the same density anomaly.

Water also has dynamic anomalies. Experiments for water show that the diffusion constant D increases on compression at low temperature T up to a maximum $D_{\max}(T)$ at $P = P_{D\max}(T)$. The behavior of normal liquids, with D decreasing on compression, is restored in water only at high P, e.g., for $P > P_{D\max} \approx 1.1$ kbars at $10 \, ^{\circ} \text{C}$. Numerical simulations for SPC/E water 12 recover the experimental results and show that the anomalous behavior of D extends to the metastable liquid phase of water at negative pressure, a region that is difficult to access for experiments. $^{13-16}$ In this region, the diffusivity D decreases for decreasing P until it reaches a minimum value $D_{\min}(T)$ at some pressure $P_{D\min}(T)$, and the normal behavior, with D increasing for decreasing P, is reestablished only for $P < P_{D\min}(T)$.

In order to shed some light in the relation between the thermodynamic and dynamic anomalies, Errington and Debenedetti investigated the pressure dependence of the structure of SPC/E water by measuring (i) the tendency of pairs of molecules to adopt preferential separations by means of a translational order parameter, and (ii) the extent to which a molecule and its four nearest neighbors assume a tetrahedral arrangement by means of an orientational order parameter.¹⁴ They found a region of the P-T phase diagram, called the structural anomalous region, where both order parameters decrease for increasing density, i.e., the liquid is more disordered at higher density, in contrast with the behavior of normal liquids. They showed that the structural anomalous region encompasses the region where the diffusion is anomalous, and that the latter includes the region where the density is anomalous. Next, Shell et al. used the two structural order parameters to study the case of a computer model for silica. Both order parameters demonstrated to be anomalous, but the hierarchy of anomalies for silica is different compared to water. For silica, the diffusion anomalous region is the wider one, containing the structural anomalous region which, in turn, has the density anomaly region inside.

The study of core-softened potentials generates models that are computationally (sometimes even analytically) tractable and that may retain some qualitative features of network forming fluids such as water and silica. What they have in common with the water potential is the two interaction scales that in the case of water are the van der Waals and the hydrogen bond interaction distances. In some cases, theoretical and experimental fersults for core-softened potentials show the presence of anomalies. In other cases, in

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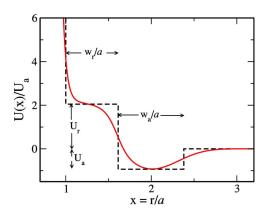


FIG. 1. (Color online) Interaction potentials studied in this work. Dashed line represents the discontinuous shouldered well (DSW) potential (Refs. 27–30). Continuous line represents the continuous shouldered well (CSW) potential introduced in Ref. 25. The parameters are explained in the text.

particular, for the discontinuous shouldered well (DSW) potential, no density anomaly is observed.^{27–30} The DSW potential consists of a hard core, a square repulsive shoulder, and an attractive square well (dashed line in Fig. 1).

In this paper, we investigate the presence of waterlike anomalies in a continuous shouldered well (CSW) potential introduced by Franzese²⁵ (continuous line in Fig. 1). Although the model is similar to the DSW version, we find density, diffusion, and structural anomalies and we observe that they occur with the same hierarchy as in water. In order to get a better understanding of the reasons why the CSW potential displays waterlike anomalies, while the DSW one does not, we apply recent ideas that have emerged on the connection between the anomalies and the excess entropy. ^{8,15,31}

Since the model studied here resembles effective potentials for complex liquids or liquid metals, our results suggest that the waterlike hierarchy of anomalies could be experimentally found in systems where pressure-induced bond shortening, rather than the change in the coordination number, is the origin of the density anomaly.²⁶

The outline of the paper is as follows. We present the details of the model in Sec. II, and the details of the simulations in Sec. III. In Sec. IV we show and discuss the results obtained from simulations and in Sec. V, we summarize and give the conclusions.

II. THE MODEL

The CSW model²⁵ studied here is composed by a system of identical particles interacting through the potential (Fig. 1) defined as

$$U(r) = \frac{U_R}{1 + \exp[\Delta(r - R_R)/a]} - U_A$$

$$\times \exp\left[-\frac{(r - R_A)^2}{2\delta_A^2}\right] + \left(\frac{a}{r}\right)^{24}, \tag{1}$$

where U_R and U_A are the energy of the repulsive shoulder and of the attractive well, respectively, a is the diameter of particles, R_R and R_A are the repulsive average radius and the distance of the attractive minimum, respectively, Δ is a parameter related to the slope of the potential at R_R , and δ_A^2 is the variance of the Gaussian centered at R_A . The many parameters of this potential allow it to assume several forms, similar to the potential proposed by Cho et al., 32 ranging from a deep double wells^{32–34} to repulsive ramplike potentials,^{20,21,35} making this equation malleable for studying different isotropic models for systems such as colloids or liquid metals. Details on the role played by these parameters may be found in Ref. 25. The set of parameters used here is the same as in Ref. 25: $U_R/U_A = 2$, $\Delta = 15$, $R_R/a = 1.6$, R_A/a =2, $(\delta_A/a)^2$ =0.1, with cutoff r_c =3.0a. To have a continuous function at the cutoff distance, a constant C/U_A =0.208 876 and a linear term $\lambda r/a$, with $\lambda/a = -0.0673794$, are added.²⁵ The CSW potential may be interpreted as a continuous version of the DSW potential investigated in Refs. 27–30 and 36. The corresponding DSW potential is represented by a square shoulder of size w_R/a and a well of size w_A/a and depth $U_R + U_A$ illustrated by the dashed line in Fig. 1 (the parameters in the figure are $w_R/a=0.612$, $w_A/a=0.767$, and $U_R/U_A=2.2$).

III. DETAILS OF SIMULATIONS

In order to investigate the interparticle CSW potential Eq. (1), we employ a molecular dynamics (MD) technique.³⁷ We consider 1000 particles in a cubic box with periodic boundary condition. We perform simulations in the *NVT* ensemble with the Nosé-Hoover thermostat³⁷ and coupling parameter equal to 2. Pressure, temperature, density, and diffusion are calculated in dimensionless units; $P^* \equiv Pa^3/U_A$, $T^* \equiv k_BT/U_A$, $\rho^* \equiv \rho a^3$, and $D^* \equiv D(m/a^2U_A)^{1/2}$.

Positions, velocities, and configurational energy of the particles are stored for every 100 steps, during a total time of 10^6 steps, given a total of 10^4 stored configurations. Pressure, diffusion, and order parameters are calculated over these configurations, discarding the first 2×10^3 for equilibration purposes. The time step used is 2×10^{-3} in units of $(a^2m/U_A)^{1/2}$ (of the order of $\approx 1.3\times 10^{-12}\,\mathrm{s}$ for argonlike atoms and $\approx 1.1\times 10^{-12}\,\mathrm{s}$ for waterlike molecules).

The model displays²⁵ a phase diagram with two first-order phase transitions ending in critical points: One transition between gas and low-density liquid (LDL) phases and another between LDL and high-density liquid (HDL) phase. The LDL-HDL phase transition is metastable with respect to the crystal phase.

For temperatures and densities inside the metastable region, the positions, velocities, and configuration energy are stored for every 10 steps and the quantities are calculated with the configurations stored until the drop in configuration energy and virial, discarding the equilibration time. The time before the drop in energy and virial characterizes the lifetime of the metastable phase.

IV. RESULTS AND DISCUSSION

A. Density anomaly

In the vicinity of the LDL-HDL critical point, the system shows density anomaly. The isochores have a temperature of minimum pressure in the interval $0.185 < \rho^* < 0.215$, corresponding to the temperature of maximum density (TMD)

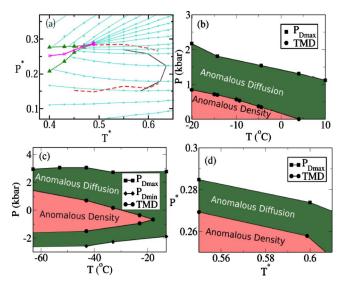


FIG. 2. (Color online) (a) Pressure-temperature (P-T) diagram for the CSW model. Lines correspond to isochores with, from bottom to top, $\rho^* = 0.175$, 0.18, 0.185, 0.19, 0.2, 0.21, 0.215, 0.22, 0.23, 0.24, 0.25, 0.26, 0.27, 0.28, 0.29, 0.3, 0.31, and 0.32. Triangles represent spinodal lines for the LDL-HDL phase transition, with LDL at low P and HDL at high P. We estimate the LDL-HDL critical point where the spinodal lines converge (large filled circle). Line with crosses represents our estimate of the liquid-liquid coexistence line. The TMD (bold continuous) line bends toward the LDL-HDL critical point at high P. Dashed lines bound the region where the diffusion anomaly occurs (see Sec. IV B). (b) Experimental data for water anomalies adapted from Angell et al. (Ref. 1). Circles denote the line of temperatures of maximum density (TMD) at constant P. Squares mark where the diffusion has a maximum value with increasing P at constant T, $P_{Dmax}(T)$. (c) Simulation data for SPC/E water adapted from Netz et al. 13 Squares mark $P_{D\mathrm{max}}$ where the diffusion has a maximum value D_{max} at constant T, and diamonds mark $P_{D \min}$ for the local minima D_{\min} . Circles locate the TMD line. (d) Zoomed region from panel (a), showing good qualitative agreement between our simulations and the experiments.

at constant P [Fig. 2(a)]. Inside the TMD line the density decreases as the system is cooled at constant P. For densities in the range of $0.2 < \rho^* < 0.215$, the TMD line is negatively sloped, while it has a positive slope for $0.185 < \rho^* < 0.2$. Our results for the TMD line in the ranges of $0.55 \le T^* \le 0.61$ and $0.25 \le P^* \le 0.30$ [Fig. 2(d)] compares well, qualitatively, with the experimental data for water, showing a negative slope [Fig. 2(b)]. The data in all the supercritical regions [Fig. 2(a)] resemble the simulation results for SPC/E water, with a retracing TMD line [Fig. 2(c)]. Nevertheless, in the present case, the TMD line is at P > 0 for all its length, while for SPC/E water it extends to negative pressures.

B. Diffusion anomaly

The diffusion coefficient is calculated using the meansquare displacement averaged over different initial times,

$$\langle \Delta r(t)^2 \rangle = \langle [r(t_0 + t) - r(t_0)]^2 \rangle. \tag{2}$$

From Eq. (2), the diffusion coefficient may be obtained as follows:

$$D = \lim_{t \to \infty} \langle \Delta r(t)^2 \rangle / 6t. \tag{3}$$

While for normal fluids, diffusivity decreases monotonically with increasing density at constant temperature, for the model Eq. (1) this is the case only for temperatures T^*

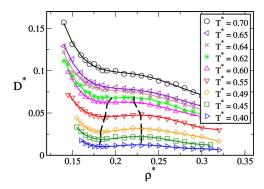


FIG. 3. (Color online) The diffusion coefficient against the density for several isotherms. For the range of densities bracketed within the dashed lines, the particles move faster under compression for temperatures lower than 0.64. This is the opposite behavior which one expects for normal fluids. Dashed lines are guides for the eyes connecting $\rho_{D_{\min}}$ and $\rho_{D_{\max}}$.

> 0.64 (Fig. 3). For T^* < 0.64, the diffusion coefficient has three regions: (i) For ρ < $\rho_{D \text{min}}$, D decreases as ρ increases as expected for normal fluids. (ii) For $\rho_{D \text{min}}$ < ρ < $\rho_{D \text{max}}$, D increases with density—an anomalous behavior. This region is called the *anomalous diffusion region*. (iii) For ρ > $\rho_{D \text{max}}$, the normal behavior is restored with D decreasing as density increases.

The line of $P_{\rm Dmax}(T)$ in the P-T phase diagram [Fig. 2(d)] is consistent with the diffusivity maxima observed in experiments for water [Fig. 2(b)]. Moreover, both lines of diffusion extrema, $P_{\rm Dmax}(T)$ and $P_{\rm Dmin}(T)$ [Fig. 2(a)], resemble the results for detailed models of water $^{13\text{-}15}$ [Fig. 2(c)], silica, 9,10,38 and other isotropic potentials. 18,20,23,24 We find that our $D_{\rm min}$, like other isotropic models, 18,20,23,24 occur at positive P.

C. Structural anomaly

The two quantities used for studying the structural behavior of the system of particles interacting through the potential in Eq. (1) are the translational order parameter t and the orientational order parameter Q_6 . The translational order parameter is defined as 9,14,39

$$t \equiv \int_0^{\xi_c} |g(\xi) - 1| d\xi, \tag{4}$$

where $\xi \equiv r \rho^{1/3}$ is the distance r in units of the mean interparticle separation $\rho^{-1/3}$, ξ_c is the cutoff distance set to half of the simulation box times $\rho^{-1/3}$, as in Ref. 21, and $g(\xi)$ is the radial distribution function proportional to the probability of finding a particle at a (reduced) distance ξ from a reference particle. For an ideal gas g=1 and t=0. In the crystal phase $g \neq 1$ over long distances and t is large.

For normal fluids, t increases with increasing density. We find this monotonic behavior only for $T^* \ge 1.8$ (Fig. 4). For $T^* < 1.8$, we observe that t has a maximum at $\rho_{tmax}(T)$, decreases for increasing $\rho > \rho_{tmax}(T)$, reaches a minimum at $\rho_{tmin}(T)$, and recovers the normal increasing behavior for $\rho > \rho_{tmin}(T)$ (Fig. 4).

For each particle $i=1,\ldots,N$, we calculate $^{9,14,39-43}$

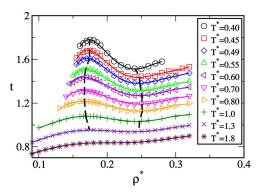


FIG. 4. (Color online) The translational order parameter as a function of density. While for normal fluids compression leads to increase the translational order parameter, for the model of Eq. (1) this is the case only for high temperatures ($T^* \ge 1.8$). Dashed lines bound the region where t behaves anomalously.

$$Q_{\ell}^{i} = \left[\frac{4\pi}{2\ell + 1} \sum_{m = -\ell}^{m = \ell} |\overline{(Y_{\ell m}^{i})_{k}}|^{2} \right]^{1/2}, \tag{5}$$

with $\ell=6$. The quantity $\overline{(Y_{\ell m}^i)}_k=1/k\sum_{j=1}^k Y_{\ell m}(\theta_{ij},\phi_{ij})$ is the average over the spherical harmonics $Y_{\ell m}$ calculated over the vectors $\mathbf{r}_{ij}(\theta_{ij},\phi_{ij})$, with $j=1,\ldots,k$, connecting particle i with its k nearest neighbors j. We use k=12 as in Ref. 21. The local orientational order 21,24 of the whole system is calculated as the average over the N particles,

$$Q_6 = \frac{1}{N} \sum_{i=1}^{N} Q_6^i. \tag{6}$$

For a crystal, Q_6 is large, while $Q_6^{\mathrm{ig}} = 1/\sqrt{k}$ for an ideal gas. Our study reveals an anomalous behavior also for the orientational order parameter Q_6 . For normal fluid, Q_6 increases monotonically with increasing density. Instead, for the present model, we find that Q_6 is nonmonotonic, displaying a maximum at $\rho_{Q\mathrm{max}}$ for all the considered temperatures (Fig. 5).

Since $\rho_{Q\max}$ lies between the densities which bound the extrema in t, $\rho_{t\max}$, and $\rho_{t\min}$, we call *structural anomalous region* the interval between $\rho_{Q\max}$ and $\rho_{t\min}$. In this region, *both* parameters t and Q_6 are anomalous, and the liquid be-

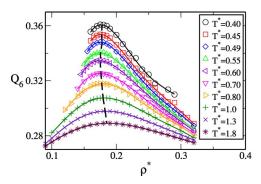


FIG. 5. (Color online) The orientational order parameter against density. We observe that Q_6 has a maximum at $\rho_{Q\text{max}}$, meaning that Q_6 decreases under compression for some range of densities. The Q_6 maxima lie between the extrema points of the translational order parameter t. Dashed line marks the location of maximum Q_6 .

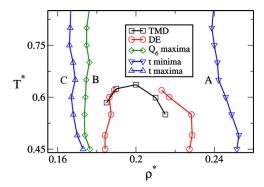


FIG. 6. (Color online) Temperature-density plane containing all the anomalies found for the CSW potential. The TMD line bounds the innermost region with the density anomaly behavior. This region is surrounded by the D extrema lines that encompass the region with diffusion anomaly. The outmost anomalous region, including the first two, is between curves B and A, where the system exhibits an anomalous behavior in structure as shown by the order parameters t and Q_6 . The curve C marks the maxima in t occurring where Q_6 has a normal behavior.

comes less ordered with increasing density, in contrast with the behavior of normal liquids.

D. The hierarchy of anomalies and order map

The relation between the several anomalies presented for the CSW potential is illustrated in Fig. 6. The TMD line lies between the diffusivity extrema (DE) lines, that are included within the structural anomalous region bounded by the curves of maxima of Q_6 and minima of t. The hierarchy of anomalies found here is the same reported for the SPC/E water and other two-scale potentials. 18,21,24

The *order map* in the t- Q_6 plane (Fig. 7) resembles the one observed for SPC/E water, ¹⁴ silica, ⁹ and other two-scale potentials ^{21,24} since it has an inaccessible region. Differently from water, and similar to other two-scale potentials, ^{21,24} the translational and orientational order parameters are not coupled into the structural anomaly region. Indeed, for the densities belonging to the structural anomalous region, t and

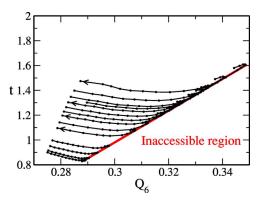


FIG. 7. (Color online) The t- Q_6 plane or order map. The arrows indicate the direction of increasing density. Each line correspond to an isotherm and from top to bottom they are T^* =0.49, 0.55, 0.60, 0.62, 0.65, 0.70, 0.75, 0.80, 0.90, 1.0, 1.3, 1.5, 1.7, and 1.8. By increasing the density, at low ρ both order parameters increase (normal fluid behavior), then at intermediate ρ they both decrease (structural anomaly region), then at higher ρ the orientational Q_6 decreases, while the translational t increases. As in the case of SPC/E water, silica, and other two-scale potentials the region with high Q_6 and low t is inaccessible. The inaccessible region is limited by a straight line t_{\min} =a+b Q_6 with a= -2.86 ± 0.02 and b= 12.8 ± 0.1 .

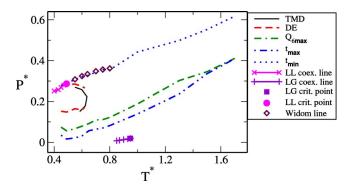


FIG. 8. (Color online) Pressure-temperature phase diagram merging all the results found for the CSW potential. The meaning of the lines is described in the legend, where DE stands for diffusivity extrema, LL for liquid-liquid, and LG for liquid-gas. See text for more details.

 Q_6 map onto a two dimensional region (Fig. 7), opposite to the case of water, in which they map onto a single line.¹⁴

E. The Widom line and the minima in t

The so-called Widom line is defined as the locus of the maximum of response functions close to the critical point at the fluid phase. Here we calculate the isothermal compressibility (Fig. 8),

$$\kappa_T = -\left(\frac{\partial \ln V}{\partial P}\right)_T. \tag{7}$$

As expected, the Widom line (diamond symbols in Fig. 8) is a continuation of the liquid-liquid coexistence line (crosses in Fig. 8). Our data show that the Widom line coincides with the minima in the translational order parameter t, at least in the vicinity of the liquid-liquid critical point. This is consistent with the observation that compressibility, structure factor g(r), and the parameter t are related. In particular, we find that κ_T has its maximum where the translational order is minimum. Hence, the largest variation of volume at constant T for increasing P occurs at the pressure where the behavior of the structural parameter t changes from anomalous to normal. To our knowledge, these are the first data showing this relation between κ_T and t.

F. Excess entropy and anomalies

Why the DSW potential has no waterlike anomalies and its continuous counterpart, the CSW potential, does? We can gain some understanding by analyzing the density dependence of the excess entropy. It has been shown⁸ that the behavior of the excess entropy as a function of density is linked to anomalies in density, diffusion, and structure.

Extrema in $(\partial S/\partial \rho)$ correspond to an thermal expansion coefficient equal to zero and, therefore, to density extrema. The thermodynamic condition that gives rise to the density anomaly can be written as $\Sigma_{\rm ex} \equiv (\partial s^{\rm ex}/\partial \ln \rho)_T > 1$, as shown by Errington *et al.* in Ref. 44. Here, $s^{\rm ex}$ is the excess entropy, defined as $s^{\rm ex} = s - s^{\rm ig}$, i.e., the difference between the entropy s of a real fluid and the entropy $s^{\rm ig}$ of an ideal gas at the same T and ρ , due to the correlations between the position of the particles of the real fluid.

Based on the proposition of Rosenfeld⁴⁵ that the logarithm of the diffusion coefficient D is proportional to the excess entropy, a nonmonotonic behavior in the excess entropy would imply a nonmonotonic behavior in D. Following the empirical Rosenfeld's parametrization,⁴⁵ Errington *et al.*⁴⁴ have also observed that the diffusion anomaly can be predicted by $\Sigma_{\rm ex} > 0.42$.

Finally, they argue that $\Sigma_{\rm ex}{>}0$ is a good estimate for determining the region where structural anomaly occurs because for normal fluids, the excess entropy decreases for increasing density at constant temperature. He cent works have explored the new possibilities one can achieve through the connections between excess entropy and structure, not only in the matter of isotropic fluids 8,31,46 but also in water, 15,47 silica, and BeF2. He had be a good estimate for determining the region where structural anomaly occurs because for normal fluids, the excess entropy decreases for increasing density at constant temperature.

To calculate the excess entropy, one should count all the accessible configurations for a real fluid and compare with the ideal gas entropy. This calculation is not straightforward and can be approximated by

$$s_2 = -2\pi\rho \int [g(r)\ln g(r) - g(r) + 1]r^2 dr,$$
 (8)

since s_2 is the dominant contribution to excess entropy $^{50-52}$ and it is proved to be between 85% and 95% of the total excess entropy in Lennard-Jones systems. 52,53 The two-body contribution s_2 depends only on the radial distribution function g(r) and the density, giving a direct connection between structure and thermodynamics. The excess entropy and the translational order parameter are linked because both depend on the deviation of g(r) from unity. The relation between excess entropy and the orientational order parameter, however, depend on the symmetries of the structures in the high and low density limits. Indeed, it turns out that the extent to which the pair correlation entropy is sensitive to orientational order determines the behavior of the structurally anomalous region.

However, as shown in Ref. 44 s_2 overestimates the excess entropy, $s^{\rm ex} < s_2$, and $\Sigma_{\rm ex} < \Sigma_2 \equiv (\partial s_2/\partial \ln \rho)_T$. Therefore, the inequalities $\Sigma_2 > 0$, $\Sigma_2 > 0.42$, and $\Sigma_2 > 1$, overestimate the region of structural, diffusion, and density anomalies, respectively. Nevertheless, Σ_2 has been shown to give estimates of the anomalous regions in qualitative agreement with the estimates based on $\Sigma_{\rm ex}$.

In order to compute Σ_2 for the DSW potential, we simulate 500 particles inside a cubic box with periodic boundary conditions. The particles interact through the DSW potential (dashed line in Fig. 1), with parameters w_R/a =0.612, w_A/a =0.767, and U_R/U_A =2.2. The equilibration and production times are 350 and 650, respectively, in reduced units. To achieve the desired temperature, we rescale the velocities until the equilibration. At equilibrium, we simulate the system in the *NVE* ensemble.

From the analysis of s_2 and Σ_2 for the DSW potential [Figs. 9(a) and 9(b)], we observe that Σ_2 has a maximum, not monotonic with T, that is always smaller than 1. Therefore, $\Sigma_{\rm ex} < 1$ and no density anomaly is expected, in agreement with Franzese *et al.*²⁷

We find densities where the DSW potential has Σ_2 > 0.42, for 0.60 \leq $T^* \leq$ 0.75, and Σ_2 > 0, for all the four tem-

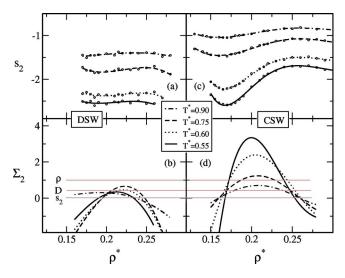


FIG. 9. (Color online) (a) Pair contribution of excess entropy s_2 for the DSW potential (dashed line in Fig. 1) against density at constant T. Circles are simulated data and lines are fifth order polynomial fit from data. (b) $\Sigma_2 = (\partial s_2/\partial \ln \rho)_T$ is shown for DSW potential. Panels (c) and (d) show the results for the CSW model. Horizontal lines mark the threshold value for anomaly in density ρ , diffusion D, and structure, as explained in the text. Solid, dotted, dashed, and dotted-dashed lines correspond to temperatures $T^* = 0.55$, 0.60, 0.75, and 0.90 in all panels.

peratures studied here [Fig. 9(b)]. These values apparently suggest the presence of diffusion and structural anomaly, respectively. However, since $\Sigma_2 > \Sigma_{\rm ex}$ and the Rosenfeld criteria holds with 30% uncertainty, ⁴⁴ our results for Σ_2 in Fig. 9(b) give no final answer about the presence of diffusion anomaly. A more detailed study, beyond the goal of this work, on diffusion and structural anomaly for the DSW potential would be necessary to clarify these points.

Our results on s_2 and Σ_2 for the CSW potential are consistent with our analysis of the anomalies described in the previous sections [Figs. 9(c) and 9(d)]. We observe that Σ_2 displays a maximum that increases for decreasing temperature. For the four studied temperatures, we find that Σ_2 >0.42 (hence, greater than zero), suggesting the occurrence of both diffusion and structural anomalies. Since we find a range of densities where $\Sigma_2 > 1$ only for $T^* \leq 0.75$, the result suggests the presence of density anomaly only below this temperature. This is qualitatively consistent with the waterlike hierarchy of anomalies that we have reported in the previous sections. However, the comparison between Figs. 9(d) and 6 shows that the criteria based on Σ_2 overestimate the temperatures and densities where the anomalies appear. We interpret this discrepancy as the effect of using Σ_2 , instead of $\Sigma_{\rm ex}$, in the criteria for the anomalies, with $\Sigma_2 > \Sigma_{\rm ex}$.

Therefore, the excess entropy analysis is a useful tool to distinguish between the DSW and the CSW potential. In the discontinuous case the increase of the excess entropy with density at constant T is not enough to give rise to the density anomaly. In the continuous case, instead, its increase is large enough at low T. For the diffusion and structural anomalies, the analysis based on the approximate expression s_2 is less clear. However, the comparison with the direct calculations of the anomalies in the CSW case suggests that only for the

TABLE I. Critical temperatures $T_{C_1}^*$ and $T_{C_2}^*$, pressures $P_{C_1}^*$ and $P_{C_2}^*$, and densities $\rho_{C_1}^*$ and $\rho_{C_2}^*$, for the gas-liquid critical point C_1 and the liquid-liquid critical point C_2 , for the CSW potential. The quantities are expressed in dimensionless units.

Critical point	T^*	P^*	$ ho^*$
Gas-liquid C_1	0.95 ± 0.06	0.019 ± 0.008	0.08 ± 0.03
Liquid-liquid C_2	0.49 ± 0.01	0.285 ± 0.007	0.247 ± 0.008

CSW potential the increase of excess entropy is enough to reach the regimes with diffusion and structural anomalies at the considered temperatures.

From Fig. 9 one observes that for the CSW potential s_2 is more sensitive to density variations than for the DSW potential when $\rho^* < 0.24$, i.e., when the average interparticle distance $r/a = (\rho^*)^{-1/3} > 1.612$. This is consistent with the fact that the DSW potential is constant for $1.612 \le r/a < 2.379$, corresponding to average density $\rho^* \le 0.24$ and $\rho^* > 0.07$, respectively. Therefore, for the DSW potential in this regime of density one can expect a small variation of g(r) and, as a consequence, of s_2 . Instead, the CSW potential for r/a > 1.612 ($\rho^* < 0.24$) has a large variation, equal to 100% of its attractive energy depth, implying a sensible variation in g(r) and s_2 .

It is worth noticing that, recently, Netz *et al.*⁵⁴ have argued that, in potentials with two characteristic scales, the region of density anomaly can be observed only if it appears at a temperature such that k_BT is larger than the discontinuity in the interaction energy (e.g., $k_BT > U_A + U_R$ in the DSW in Fig. 1). This is consistent with our results for the CSW potential, showing the density anomaly at $k_BT = 0.65U_A > 0$, because the CSW potential has no discontinuity in the interaction energy. On the other hand, for the DSW potential with discontinuity $U_A + U_R = 3.2U_A$, since no density anomaly is found for $k_BT > 3.2U_A$, this criterion excludes that the anomaly could be found at lower temperatures.

V. SUMMARY AND CONCLUSIONS

We studied a three dimensional system of particles interacting through a continuous isotropic interparticle pair potential recently proposed by Franzese. The potential has a hard core, a repulsive shoulder, and an attractive well. This CSW potential resembles the DSW potential studied in Ref. 27 and shows new features that are absent in the DSW potential.

As its discontinuous counterpart, the CSW potential displays a phase diagram with a gas-liquid phase transition ending in a critical point, and a LDL-HDL phase transition ending in a liquid-liquid critical point. Table I shows the values of the critical parameters for the two critical points. The LDL-HDL phase transition is metastable with respect to the crystal phase. In the *P-T* phase diagram the LDL-HDL phase transition line has positive slope, as in the discontinuous model and in the ramp potential. This feature suggests that these models are describing systems different from water, because for water and waterlike models, the slope of the liquid-liquid phase transition is expected to be negative. S6-59

In contrast with its discontinuous counterpart, the CSW potential has density anomaly (Fig. 8).²⁵ The TMD line bends toward the LDL-HDL critical point at high P, as also seen for other isotropic potentials.⁵⁵

We find that the CSW potential has also diffusion and structural anomalies (Fig. 8), as water and silica. Since water¹⁴ and silica⁹ have different order of these anomalies, we investigate the cascade of anomalies for the CSW potential. We find that for the CSW model the structural anomalous region encompasses the diffusion anomaly region, inside which the density anomaly region is observed, as in water. This is consistent with the previous analysis of other two-scale isotropic potentials. 21,23

The loci of structural anomalies and the two phase transition lines in the P-T phase diagram display a positive slope. The TMD line has a positive slope at low P and a negative slope at high P. By decreasing T, we find that three quantities—the minima of the structural order parameter, the Widom line, and the maxima of diffusivity—approach the LDL-HDL critical point value, suggesting that the high-T behavior of these quantities could give an indirect indication of the location of the LDL-HDL critical point in real systems, if it is present. In particular, close to the LDL-HDL critical point, the Widom line coincides with the minima of the translational structural order parameter.

To understand the difference between the DSW and the CSW potential, we perform an excess entropy analysis, approximating the excess entropy by the pair correlation entropy s_2 . Comparison of this analysis with the direct calculation of the anomalies suggests that the approximate excess entropy s₂ satisfies criteria with empirical thresholds higher then those predicted for the exact excess entropy s^{ex} .

Interestingly, the excess entropy analysis for the DSW and the CSW potential emphasizes a relevant difference for the appearance of the anomalies: The DSW potential is constant in the density regime of the anomalies (approximately for $0.17 < \rho^* < 0.25$, i.e., average interparticle distance 1.6 < r/a < 1.8), while the CSW potential has r-dependent soft repulsion. This observation could clarify also why potentials, such as the ramp in Ref. 55, show waterlike anomalies because they have appreciable distance dependence (soft repulsion) for interparticle separation between the repulsive and attractive length.

Our work confirms that waterlike anomalies can be present also in systems that, unlike water, have no directional bonds, as discussed in the previous works. 18-24,36 The CSW potential and other two-scale isotropic models 18-24,36,55 have features that are different from water, such as the positive slope in the P-T phase diagram of the LDL-HDL phase transition, or the locus of diffusion minima at positive pressure, instead of negative. 13 These differences emphasize the possible existence of liquids with anomalies induced by bond shortening at high P rather than the change in the coordination number as in water or silica. This conclusion is consistent with recent results on metallic glasses showing polyamorphism.²⁰

We conclude by remarking that, although a similar cascade of anomalies was observed in other two-scale potentials, ^{18,21,24} the CSW potential is the first with a strong shoulder that displays these anomalies. This feature makes the CSW potential suitable to study liquid metals, such as Cs, Mg, Sr, Ba, 60 and colloids, 61 where effective interactions with large shoulders are derived in some cases. By varying the parameters of Eq. (1), the potential changes from deep double wells to a repulsive ramp, and by varying the power of the core of Eq. (1)—here equal to 24 to mimic a hard core—the stiffness of the core changes, making Eq. (1) an interesting functional form to model and interpolate pressure-dependent effective interactions.

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¹C. A. Angell, E. D. Finch, and P. Bach, J. Chem. Phys. **65**, 3065 (1976).

²H. Thurn and J. Ruska, J. Non-Cryst. Solids **22**, 331 (1976).

³ Periodic Table of elements (http://periodic.lanl.gov/default.htm, 2007).

⁴G. E. Sauer and L. B. Borst, Science **158**, 1567 (1967).

⁵S. J. Kennedy and J. C. Wheeler, J. Chem. Phys. **78**, 1523 (1983).

⁶T. Tsuchiya, J. Phys. Soc. Jpn. **60**, 227 (1991).

⁷C. A. Angell, R. D. Bressel, M. Hemmatti, E. J. Sare, and J. C. Tucker, Phys. Chem. Chem. Phys. 2, 1559 (2000).

⁸R. Sharma, S. N. Chakraborty, and C. Chakravarty, J. Chem. Phys. 125, 204501 (2006).

⁹M. S. Shell, P. G. Debenedetti, and A. Z. Panagiotopoulos, Phys. Rev. E 66, 011202 (2002).

¹⁰P. H. Poole, M. Hemmati, and C. A. Angell, Phys. Rev. Lett. 79, 2281 (1997).

¹¹S. Sastry and C. A. Angell, Nat. Mater. **2**, 739 (2003).

¹²H. J. C. Berendsen, J. R. Grigera, and T. P. Straatsma, J. Phys. Chem. **91**, 6269 (1987).

¹³P. A. Netz, F. W. Starr, H. E. Stanley, and M. C. Barbosa, J. Chem. Phys. **115**, 344 (2001).

¹⁴J. R. Errington and P. D. Debenedetti, Nature (London) 409, 318 (2001).

¹⁵ J. Mittal, J. R. Errington, and T. M. Truskett, J. Phys. Chem. B 110, 18147 (2006).

¹⁶P. Kumar, G. Franzese, and H. E. Stanley, Phys. Rev. E 73, 041505 (2006).

¹⁷A. Mudi, C. Chakravarty, and R. Ramaswamy, J. Chem. Phys. 122, 104507 (2005).

¹⁸L. Xu, S. Buldyrev, C. A. Angell, and H. E. Stanley, Phys. Rev. E 74, 031108 (2006).

¹⁹ A. B. de Oliveira and M. C. Barbosa, J. Phys.: Condens. Matter **17**, 399 (2005).

²⁰ A. B. de Oliveira, P. A. Netz, T. Colla, and M. C. Barbosa, J. Chem. Phys. 124, 084505 (2006).

²¹ A. B. de Oliveira, P. A. Netz, T. Colla, and M. C. Barbosa, J. Chem. Phys. 125, 124503 (2006).

²²H. M. Gibson and N. B. Wilding, Phys. Rev. E **73**, 061507 (2006).

²³ Z. Yan, S. V. Buldyrev, N. Giovambattista, and H. E. Stanley, Phys. Rev. Lett. 95, 130604 (2005).

²⁴Z. Yan, S. V. Buldyrev, N. Giovambattista, P. G. Debenedetti, and H. E. Stanley, Phys. Rev. E 73, 051204 (2006).

²⁵G. Franzese, J. Mol. Liq. **136**, 267 (2007).

²⁶ H. W. Sheng, H. Z. Liu, Y. Q. Cheng, J. Wen, P. L. Lee, W. K. Luo, S. D. Shastri, and E. Ma, Nat. Mater. 6, 192 (2007).

²⁷G. Franzese, G. Malescio, A. Skibinsky, S. V. Buldyrev, and H. E. Stanley, Nature (London) 409, 692 (2001).

²⁸ G. Franzese, G. Malescio, A. Skibinsky, S. V. Buldyrev, and H. E. Stanley, Phys. Rev. E 66, 051206 (2002).

²⁹ A. Skibinsky, S. V. Buldyrev, G. Franzese, G. Malescio, and H. E. Stanley, Phys. Rev. E 69, 061206 (2005).

- ³⁰G. Malescio, G. Franzese, A. Skibinsky, S. V. Buldyrev, and H. E. Stanley, Phys. Rev. E 71, 061504 (2005).
- ³¹ J. Mittal, J. R. Errington, and T. M. Truskett, J. Chem. Phys. **125**, 076102 (2006).
- ³²C. H. Cho, S. Singh, and G. W. Robinson, Phys. Rev. Lett. **76**, 1651
- ³³C. H. Cho, S. Singh, and G. W. Robinson, J. Chem. Phys. **107**, 7979 (1997).
- ³⁴P. A. Netz, J. F. Raymundi, A. S. Camera, and M. C. Barbosa, Physica A **342**, 48 (2004).
- ³⁵ A. B. de Oliveira, M. C. Barbosa, and P. A. Netz, Physica A 386, 744 (2007).
- ³⁶ A. Balladares and M. C. Barbosa, J. Phys.: Condens. Matter **16**, 8811 (2004).
- ³⁷D. Frenkel and B. Smit, *Understanding Molecular Simulation*, 1st ed. (Academic, San Diego, 1996).
- ³⁸R. Sharma, A. Mudi, and C. Chakravarty, J. Chem. Phys. 125, 044705
- ³⁹ J. E. Errington, P. G. Debenedetti, and S. Torquato, J. Chem. Phys. 118, 2256 (2003).
- ⁴⁰P. J. Steinhardt, D. R. Nelson, and M. Ronchetti, Phys. Rev. B 28, 784
- ⁴¹S. Torquato, T. M. Truskett, and P. G. Debenedetti, Phys. Rev. Lett. **84**, 2064 (2000).
- ⁴²T. M. Truskett, S. Torquato, and P. G. Debenedetti, Phys. Rev. E **62**, 993
- ⁴³ A. Huerta, G. G. Naumis, D. T. Wasan, D. J. Henderson, and A. D. Trokhymchuk, J. Chem. Phys. 120, 1506 (2004).

- ⁴⁴ J. R. Errington, T. M. Truskett, and J. Mittal, J. Chem. Phys. **125**, 244502
- ⁴⁵ Y. Rosenfeld, J. Phys.: Condens. Matter **11**, 5415 (1999).
- ⁴⁶S. N. Chakraborty and C. Chakravarty, Phys. Rev. E **76**, 011201 (2007).
- ⁴⁷ A. Scala, F. W. Starr, E. La Nave, F. Sciortino, and H. E. Stanley, Nature (London) 406, 166 (2000).
- ⁴⁸ M. Agarwal, R. Sharma, and C. Chakravarty, J. Chem. Phys. 127, 164502 (2007).
- ⁴⁹ M. Agarwal and C. Chakravarty, J. Phys. Chem. B **111**, 13294 (2007).
- ⁵⁰H. S. Green, The Molecular Theory of Fluids (North-Holland, Amsterdam, 1952).
- ⁵¹ H. J. Raveché, J. Chem. Phys. **55**, 2242 (1971).
- ⁵² A. Baranyai and D. J. Evans, Phys. Rev. A **40**, 3817 (1989).
- ⁵³S. N. Chakraborty and C. Chakravarty, J. Chem. Phys. **124**, 014507
- ⁵⁴P. A. Netz, S. Buldyrev, M. C. Barbosa, and H. E. Stanley, Phys. Rev. E 73, 061504 (2006).
- ⁵⁵ E. A. Jagla, Phys. Rev. E **63**, 061501 (2001).
- ⁵⁶P. H. Poole, F. Sciortino, U. Essmann, and H. E. Stanley, Nature (London) 360, 324 (1992).
- ⁵⁷I. Brovchenko, A. Geiger, and A. Oleinikova, J. Chem. Phys. **123**, 044515 (2005).
- ⁵⁸G. Franzese, M. I. Marques, and H. E. Stanley, Phys. Rev. E **67**, 011103
- ⁵⁹ F. Saika-Voivod, F. Sciortino, and P. H. Poole, Phys. Rev. E **63**, 011202
- ⁶⁰ J. F. Wax, R. Albaki, and J. L. Bretonnet, Phys. Rev. B **62**, 14818 (2000).
- ⁶¹M. Quesada-Perez, A. Moncho-Jorda, F. Martinez-Lopez, and R. Hidalgo-Álvarez, J. Chem. Phys. 115, 10897 (2001).