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Citation: *Applied Physics Letters* **63**, 105 (1993); doi: 10.1063/1.109727

View online: <http://dx.doi.org/10.1063/1.109727>

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Low-temperature ion-induced epitaxial growth of α -FeSi₂ and cubic FeSi₂ in Si

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(Received 22 December 1992; accepted for publication 25 March 1993)

Ion-beam-induced epitaxial crystallization of amorphous Si implanted with Fe to 18 at. % peak concentration was studied. The structure of the specimen was characterized using transmission electron microscopy and Rutherford backscattering spectrometry. Both cubic FeSi₂ and α -FeSi₂ were formed in epitaxy with the Si matrix with two types of orientations (fully aligned and twinned). The twins of α -FeSi₂ and those of cubic FeSi₂ were found to have exactly the same type of epitaxial relationship as for the aligned ones. The thermodynamically stable β -FeSi₂ is not formed, demonstrating that ion-beam-induced crystallization can lead to preferential phase formation as well as to epitaxy.

To date, much effort has been devoted to the study of the epitaxy of β -FeSi₂ on Si because it is a semiconductor that exhibits a direct transition at a band gap of ≈ 0.9 eV and thus of great interest for optoelectronic devices.¹⁻³ Only limited work, however, has been reported for the other two FeSi₂ phases which are very different from β -FeSi₂ in many aspects. α -FeSi₂ is a metallic tetragonal phase stable above ≈ 950 °C, as opposed to β -FeSi₂ which is orthorhombic and stable at temperatures below ≈ 950 °C.^{3,4} The newly discovered cubic FeSi₂ phase^{2,5-10} is metallic and, probably, magnetic^{5,10} with a fluorite structure ($a_c \approx a_{Si} = 0.5431$ nm) that can be produced only under restricted conditions. Crystallographic analysis indicates that the α phase is better lattice matched with Si than the β phase, while the cubic phase is best matched with Si among the three FeSi₂ phases, because of $a_c \approx a_{Si}$. In the case of α -FeSi₂ ($a_\alpha = b_\alpha = 0.2695$ nm and $c_\alpha = 0.514$ nm), since $2a_\alpha = 0.5390$ nm, the lattice mismatch between α -FeSi₂ and Si or cubic FeSi₂ is only 0.7% along the a or b axis, as compared to a mismatch of $\approx 2\%$ for β -FeSi₂ grown epitaxially on a Si substrate.¹¹

The first reported cubic FeSi₂ phase was obtained either by molecular beam epitaxy (MBE) combined with solid phase epitaxy⁵ or by reactive deposition epitaxy⁶ on a Si(111) substrate. This phase is metastable and transforms into β -FeSi₂ upon annealing. The transition temperature was found to decrease with increasing thickness of the epilayer.⁵ For a 30-nm-thick layer, the transition temperature is ≈ 200 °C. This result indicates that the interface energy between the epilayer and the substrate plays a crucial role in stabilizing the cubic FeSi₂ phase. More recently, we have

reported the formation of cubic FeSi₂ in an Fe-implanted Si by an ion-beam-induced epitaxial crystallization (IBIEC) process.⁹ In that case, the gain in interface energy is larger, because the cubic FeSi₂ is formed as small particles (diameter ≈ 5 nm) buried in the Si matrix and thus has a greater interface-to-volume ratio than a thin film grown on Si. The cubic phase precipitates were found to be stable above 520 °C. Moreover, two types of precipitates were observed, i.e., A-type and B-type precipitates, corresponding to full alignment of cubic FeSi₂ with the Si matrix and twinning along {111} of A type, respectively. In this letter, we present a study of Si implanted with Fe to a high dose (≈ 18 at. %), using transmission electron microscopy (TEM) and Rutherford backscattering spectrometry combined with channeling (RBS/C). It demonstrates, for the first time, not only the epitaxial formation of the high-temperature α -FeSi₂ phase by IBIEC at a low temperature (320 °C), but also the coexistence of this phase with cubic FeSi₂ in Si.

Si(001) wafers (p type, 1 Ω cm) were implanted with 50-keV Fe⁺ ions to a fluence of 4×10^{16} cm⁻² (≈ 18 at. % at the peak concentration). During implantation, the specimens were maintained at room temperature and channelled implantation was avoided. The Fe-implanted amorphous layer was crystallized by IBIEC, i.e., by irradiating the specimens at 320 °C with 500-keV Si⁺ ions to a fluence of 7.2×10^{16} cm⁻²; the beam flux was 1 μ A cm⁻².

The structure was characterized using cross-sectional TEM, including selected-area diffraction (SAD). TEM specimens were prepared by mechanical polishing and ion milling. The same specimens were also studied by RBS/C with a 1.2-MeV ⁴He⁺ beam aligned randomly and the channeling direction (Si [001]), using a standard goniometer arrangement.

The RBS/C spectra (not shown) from Si indicate that

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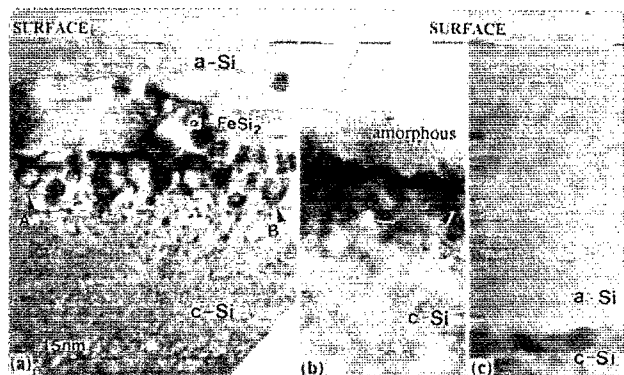


FIG. 1. Cross-sectional TEM micrographs of Si implanted at room temperature with 18 at. % Fe peak concentration, showing the phase distribution after different treatments at 320 °C: (a) After Si⁺ irradiation to $7.2 \times 10^{16} \text{ cm}^{-2}$. Under the α -FeSi₂ layer are aligned (A-type) and twin-oriented (B-type) cubic FeSi₂ precipitates, as indicated by arrows A and B. (b) After Si⁺ irradiation to $3.6 \times 10^{10} \text{ cm}^{-2}$. Below the amorphous top layer are cubic FeSi₂ precipitates, as indicated by arrows. (The weak contrast of the precipitates is because they are in a relatively thick area.) (c) After thermal annealing only. The thick layer above c-Si is completely amorphous.

the initial Fe implantation produces an amorphous Si (*a*-Si) layer ≈ 100 -nm thick and that the subsequent ion irradiation promotes crystallization and reduces this *a*-Si layer by ≈ 40 nm. On the other hand, the Fe peak indicates the formation of an Fe-containing region ≈ 35 -nm wide and a Gaussian-like distribution of Fe atoms. The IBIEC process causes only a slight narrowing of the Fe peak; neither peak shift nor channeling effect is observed.

TEM observation of the same ion-irradiated specimen reveals a layered structure. The TEM image in Fig. 1(a) shows that the specimen consists of a surface amorphous Si layer (*a*-Si) ≈ 15 nm thick¹² above an α -FeSi₂ layer (≈ 20 nm thick) and a crystalline Si (*c*-Si) region which contains spherical-shaped cubic FeSi₂ precipitates over a width of ≈ 40 nm. These cubic FeSi₂ particles decrease in size and, at the same time, increase in number density with increasing depth. The typical diameter of the cubic FeSi₂ particles is ≈ 10 nm adjacent to the interface with α -FeSi₂, but rapidly decreases to ≈ 1 nm at a greater depth. The A-type and B-type cubic FeSi₂ precipitates, previously found in a Si implanted to a low Fe dose (≈ 5 at. %),⁹ are also observed [indicated by the arrows in Fig. 1(a)]. Note that, while the large particles adjacent to the lower interface of α -FeSi₂ are of both A type and B type, the small particles in the lower region are predominantly of A type.

The phase distribution in the specimen is related to the Fe concentration profile. Comparison between the TEM and the RBS/C data shows that the α -FeSi₂ layer is located around the Fe profile peak, whereas the cubic FeSi₂ phase region corresponds to the tail of the Fe profile. This relationship leads to three observations.

First, since cubic FeSi₂ precipitates are only found at the profile tail, their formation must be associated with a low Fe concentration, in agreement with our previous results.⁹

Second, the size and density of cubic FeSi₂ precipitates

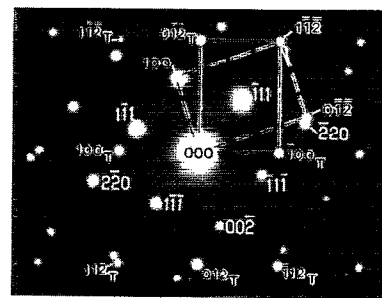


FIG. 2. A SAD pattern taken from a lower interface area of the α -FeSi₂ layer shown in Fig. 1(a). The diffraction spots correspond to the reflections from the [110] zone axis of Si or A-type cubic FeSi₂ (white indices) and from the [021] zone axis of two α -FeSi₂ crystals (outlined black indices) twinned along (11 $\bar{2}$). The spots indicated by the dashed and the solid lines correspond to an α -FeSi₂ grain in epitaxy with Si and its twin (indices with subscript *T*), respectively.

vary as a function of Fe concentration, i.e., the decrease of Fe concentration at the profile tail results in a decrease in size and an increase in density of the precipitates. Moreover, their orientation tends to change from A type to B type, as the Fe concentration (hence the precipitate size) increases. This implies that for the small particles (< 3 nm in diameter), A-type orientation is more stable than B type.

Third, the formation of the α -FeSi₂ layer around the peak of the Fe profile suggests that there exists a size limit for the growth of cubic FeSi₂ beyond which the cubic phase becomes unstable and a new phase grows in lieu of cubic FeSi₂. The existence of such a stability limit was already apparent for cubic FeSi₂ epilayers grown on Si(111) surface by MBE and the stable phase of the epilayers was β -FeSi₂, in agreement with the equilibrium phase diagram.⁵ The major difference here is the absence of the β phase and the formation of the α phase at a temperature ≈ 600 °C below its equilibrium stability limit.

The α -FeSi₂ layer consists of grains with a diameter ranging from 20 to 100 nm. Some are in epitaxy with the underlying Si or A-type cubic FeSi₂ and others are twinned with respect to the epitaxial one along the α -FeSi₂{112} planes. Figure 2(a) shows a typical SAD pattern taken from an area including α -FeSi₂, Si, and cubic FeSi₂. From this pattern the orientation relationship between the epitaxial α -FeSi₂ grains and Si (or A-type cubic FeSi₂) is determined as α -FeSi₂ [021] \parallel Si[110] and α -FeSi₂ (100) \parallel Si(002). This relationship implies that the *b* and *c* axes of the α -FeSi₂ unit cell are lying in a plane parallel to Si (001), i.e., the lower interface of the α -FeSi₂ layer, and are virtually parallel to the *b* and *a* axes of Si (or A-type cubic FeSi₂), respectively, because the angle between the *b_α* and *b_{Si}* axes (or *c_α* and *a_{Si}*) is only 1.3°. Thus, the lattice mismatch in the interface is $\approx 0.7\%$ and 5% along α -FeSi₂ [010] and [001], respectively. A slight deviation from the above epitaxial relationship, i.e., a relative rotation of two parallel planes by an angle (θ) up to $\approx 3^\circ$, is often observed, such as occurs for α -FeSi₂ (0 $\bar{1}$ 2) with Si($\bar{2}$ 20) ($\theta \approx 3^\circ$) in Fig. 2. This rotation explains the absence of channeling for the Fe peak in the RBS/C spectra.

In Fig. 2, one also finds the diffraction spots from the α -FeSi₂ grains twinned along (11 $\bar{2}$). We find that all the twin-oriented grains of α -FeSi₂ layer have the same type of orientation relationship as shown above with respect to twin-oriented cubic FeSi₂ particles (B type). In fact, since $2a_\alpha \approx a_c$ and $c_\alpha \approx a_c$, the twin planes {112} of α -FeSi₂ are equivalent to the twin planes {111} of cubic FeSi₂. This indicates that the twinned α -FeSi₂ phase is formed in epitaxy with the B-type cubic FeSi₂, same as the epitaxial α -FeSi₂ phase with the A-type cubic FeSi₂.

In order to check that the α and cubic phases are selectively formed in the IBIEC process, we performed two complementary experiments, in which two samples were prepared under conditions identical to those described above, except as regards the following. In one case, IBIEC with $3.6 \times 10^{16} \text{ Si}^+ \text{ cm}^{-2}$ was performed in order to recrystallize roughly half the as-implanted amorphous layer. No FeSi₂ precipitate is found in the remaining amorphous layer ($\approx 35 \text{ nm}$ thick) [Fig. 1(b)], proving that FeSi₂ precipitates are formed only when the Si is recrystallized. In the other case, no IBIEC was performed, but the sample was annealed at 320 °C for 3.2 h, i.e., a duration identical to that of the IBIEC procedure with $7.2 \times 10^{16} \text{ Si}^+ \text{ cm}^{-2}$. No thermally induced crystallization occurs at all (amorphous layer $\approx 95 \text{ nm}$) [Fig. 1(c)], indicating that IBIEC is indispensable for FeSi₂ precipitation at 320 °C. These two experiments also prove that no precipitation occurs at all during Fe implantation at room temperature.

From the thermodynamic point of view, the formation of α -FeSi₂ at 320 °C by IBIEC is surprising, because this phase is stable only at high temperatures ($> \approx 950 \text{ °C}$), as opposed to the low-temperature β -FeSi₂ phase which is stable below $\approx 950 \text{ °C}$. The mechanism of the α -FeSi₂ phase formation warrants discussion. IBIEC is a nonequilibrium process for which kinetic factors may be important. This process involves epitaxial growth of a crystal with the crystalline-amorphous (c - a) interface migrating toward the surface at a speed depending on irradiation flux and temperature.¹³ In our case, the crystallization speed of Si is measured as $\approx 1.5 \text{ nm min}^{-1}$. Obviously, in this crystallization process, epitaxial growth of Si is favored, because it involves a lower interfacial energy than growth in a random orientation. For the same reason, when FeSi₂ is involved in the crystallization process, not only its orientation, but also its phase, may be determined by the underlying crystal structure, especially when the precipitate dimension is small and the interfacial energy makes a significant contribution to the total free energy of the system.

As the c - a interface moves toward the surface, the cubic FeSi₂ phase is first formed at the tail of the Fe profile, because it is small in size and the interfacial energy is minimized by matching with the Si matrix, as in the case of a thin epilayer grown on a Si substrate.⁵⁻⁷ As the c - a in-

terface approaches the peak of the Fe profile, the size of the cubic precipitates increases rather rapidly to such a point that the decrease in interfacial energy by lattice matching with Si is insufficient to maintain the stability of the cubic phase and, consequently, a phase with a lower volume free energy, and possibly also a lower strain energy begins to grow. If we assume that the interfacial energy is also the controlling factor for nucleation of this new phase, α -FeSi₂, rather than β -FeSi₂, should be formed, because it has a better lattice match with Si or cubic FeSi₂ than β -FeSi₂. The strongest evidence in support of this epitaxial growth mechanism is the fact that all the grains in the α -FeSi₂ layer are found to be in perfect alignment with either the A-type or the B-type cubic FeSi₂ with exactly the same orientation relationship.

We are grateful for the technical assistance provided by the Semiramis Group (Orsay). The electron microscopy studies reported in this work were supported by the Director, Office of Energy Research, U.S. Department of Energy under Contract No. DE-AC0376 SF00098. We wish to thank W. Swider for her help in TEM specimen preparation, and C. Clerc, F. Fortuna, M. O. Ruault, J. Chevrier, and J. Derrien for valuable comments.

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